

LiCu₂O₂,

01.04.07

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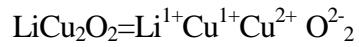
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[1].

[2],

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LCO

LCO

LCO,

LiCu_2O_2 ,

LiCu₂O₂,
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LiCu₂O₂

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1. Li₂CuO₂–CuO_x;LiCu₂O₂Li(Cu,Zn)₂O₂, (Li,Ag)Cu₂O₂;

4 10 10 ;

LiCu₂O₂.2. Li(Cu_{1-x}Zn_x)₂O₂, (Li_{1-x}Ag_x)Cu₂O₂

= 0 – 0,12 = 0 – 0,04,

Zn Ag

LiCu₂O₂Li(Cu_{1-x}Zn_x)₂O₂, (Li_{1-x}Ag_x)Cu₂O₂.

3.

σ_{DC},() = σ_{AC}LiCu₂O₂ Li(Cu,Zn)₂O₂, (Li,Ag)Cu₂O₂ 4,2 – 300 0,1 – 100 .

3 .

, LiCu₂O₂Li(Cu,Zn)₂O₂,(Li,Ag)Cu₂O₂

DC T ~ 300 K

- (DC = σ₀exp(E_a/k_BT))(), DC = Aexp(T₀/T)^{1/4}.

~25 DC

3. LiCu_2O_2
- $a, b = c: \dagger_a : \dagger_b : \dagger_c = 2 : 1 : 10^4$.
3. $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2, (\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2, (x(\text{Zn}) = 0,05, x(\text{Ag}) < 0,02)$
- S-
4. LCO
- , ,
LCO , ,
, , ,
1 - 2 ,
150 K. , ,
, ,
LiCu₂O₂
Li(Cu,Zn)₂O₂, (Li,Ag)Cu₂O₂
:
- , ,
, ,
LiCu₂O₂
,
- 4- : . « (11),

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59 5 .

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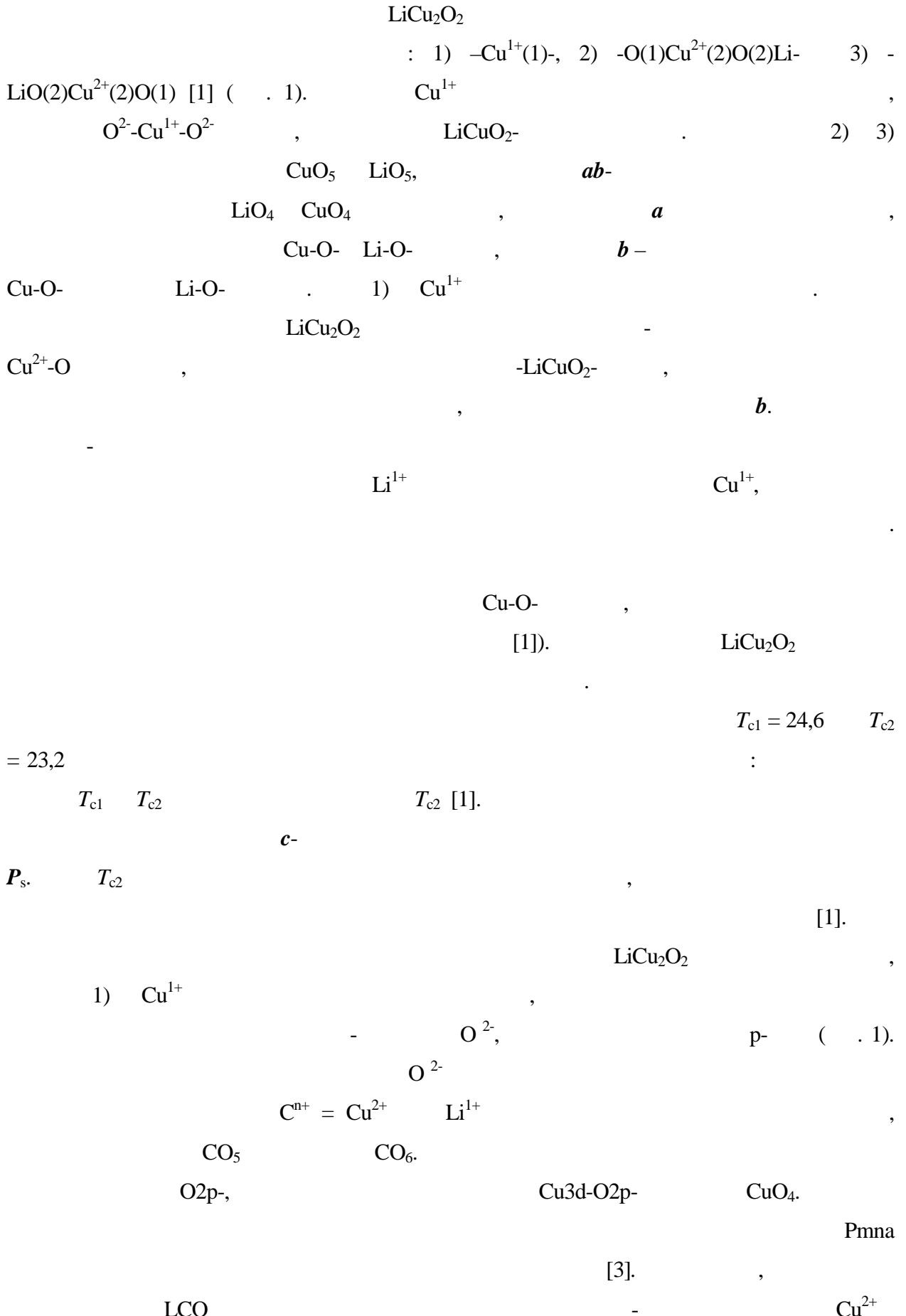
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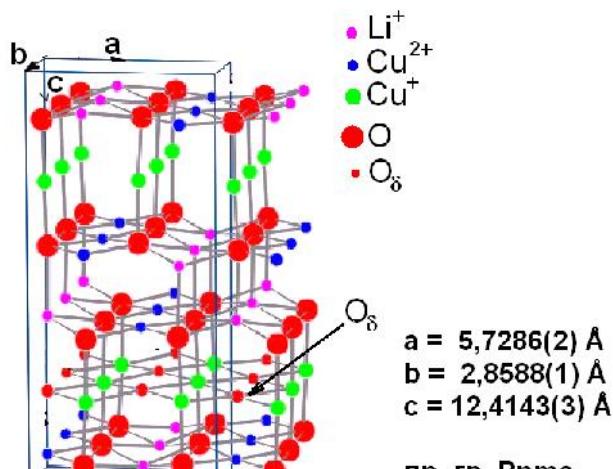
—————

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LiCu₂O₂

CuO₄-
(ladder compound),





1.

 LiCu_2O_2

[1]).

 Li^+

u(Li)-O-Cu(Li)

LCO.

 LiCu_2O_2

,

 LiCu_2O_2 (Li,Na) Cu_2O_2 [4], $\text{Li}(\text{Cu},\text{Zn})_2\text{O}_2$ $\text{Li}(\text{Cu},\text{Ni})_2\text{O}_2$ [5]. LiCu_2O_2

,

,

: 1)

2)

-2-

().

()

-4

: $(\text{CuK}_1) = 1,54051$, $(\text{CuK}_2) = 1,54433$, $(\text{CuK}_{\text{av}}) = 1,54178 \text{ \AA}$.

“Orbis” “EDAX” () Si(Li)
 , , Na.
 () Q1500
 D F. Paulik, J. Paulic, L. Erdey.

7-20 100 – 100 ,
 ,
 ,
 R_l = 5,26 ,
 5 – 300 SQUID
 MPMS-XL-7 Quantum Design Inc (H = 20)
 (zero- field-cooled – ZFC) (field-cooled – FC).
 _____ , ,
 LiCu₂O₂ ,

3.1.

Li₂CuO₂ - CuO

(. 2).

: LiCu₂O₂ LiCu₃O₃,

(. 1323 1373) (1163 1113
).

1173 - 1323

LiCu₂O₂.

CuO « . ». Li₂CO₃ « »..

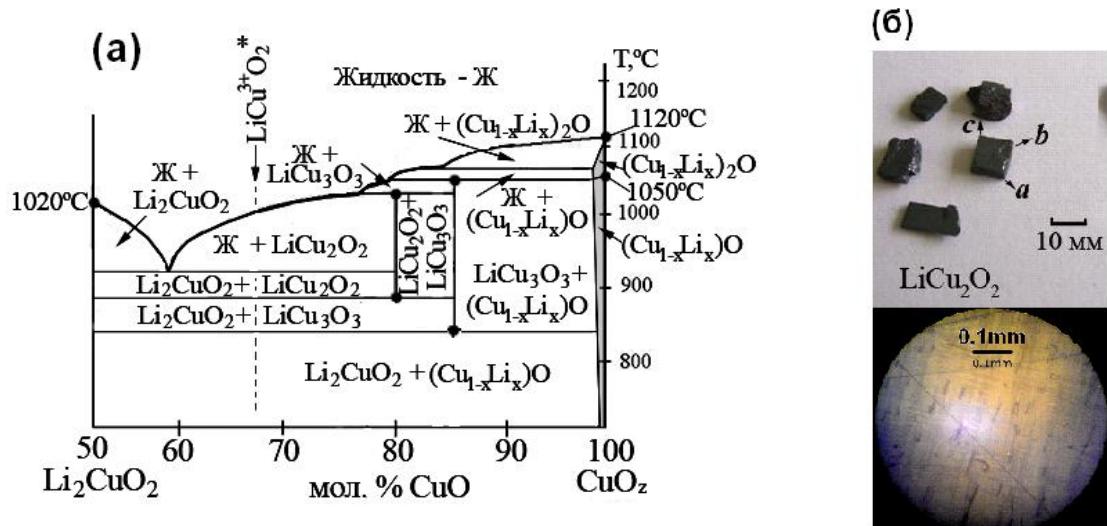
LiCu₂O₂ xCuO(1-x)Li₂CO₃ c 0,77 x 0,83 4 1393 0,5

, . 1323 ,

2,0 / - 1173 ,

1173 10 - 20

,



$$\begin{array}{ll}
 2.) & \text{Li}_2\text{CuO}_2 - \text{CuO}_z \\
 ;) & \text{LiCu}_2\text{O}_2 (\quad) \\
 \text{LiCu}_2\text{O}_2, & (\quad) \quad (001).
 \end{array}$$

1393 1323 ,

~ 1173 K LiCu_2O_2

$$\text{LiCu}_2\text{O}_2 \quad , \quad (0.5 - 4) \ 8 \ 8 \quad ^3 \ (\quad . \ 2).$$

(001),

(210) (2-10),

{001} {210}.

LiCu₂O₂,

(120) (1-20) (. 2).

LiCu₂O₂

, LCO,
, , ,
 $>0,15$ 1 .

LiCu₂O₂.

LiCu₂O₂,

6

90

5,0

6

20

a

,

LiCu₂O₂

~1113

40

1113

400

1 - 4 .

LiCu₂O₂

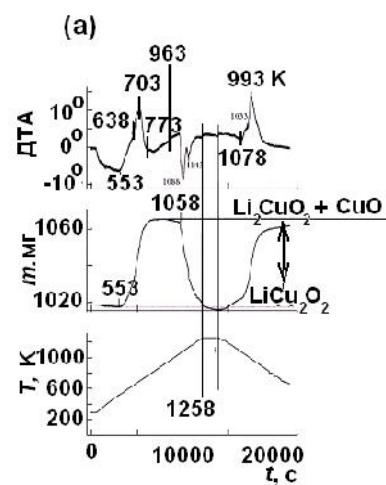
LiCu₂O₂

[3],

$T = 553\text{--}773$

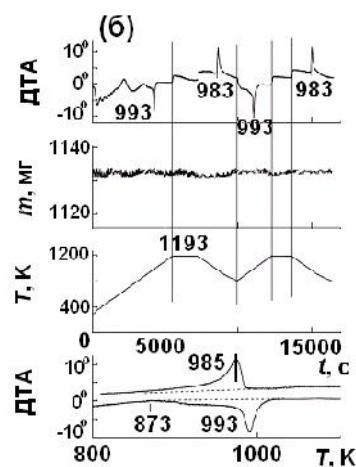
LiCu₂O₂ Li₂CuO₂ CuO.

m



3.

)



LiCu₂O₂:)

(. . 3)	Cu^+	Cu^{2+} ,
$\text{LiCu}^{2+}\text{Cu}^+\text{O}_2$.	.
1073–1223	.	LiCu_3O_3 ,
LiCu_2O_2 ,	.	.
$\text{Li}(\text{Cu}_{0.95}\text{Zn}_{0.05})_2\text{O}_2$.	LiCu_2O_2 .

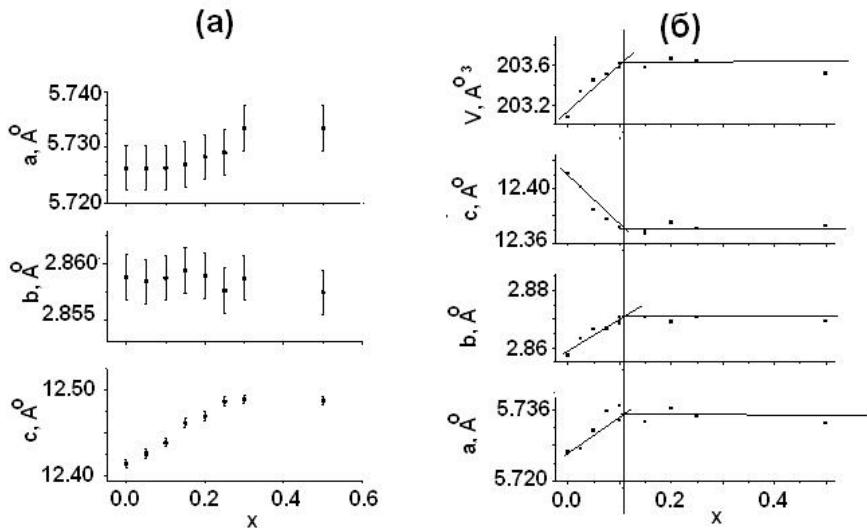
LiCu_2O_2					
1	-		2	400	
2	-		-		
1			-		
2			-		
A4	-	40 . . . 1113	2 . . . 4	400	
W2	-	40 . . . 1113			

 LiCu_2O_2 LiCu_2O_2 (. . 1320). $T_{\text{PT}} = 993$

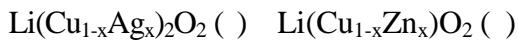
(. . 3).

 $T = 983$ LiCu_2O_2 $a = 5,73, b = 2,86, c = 12,41 \text{ \AA},$ [1] LiCu_2O_2 . ZnO , Zn LiCu_2O_2 ((Li,Zn)Cu₂O₂ - I, $\text{Li}(\text{Cu},\text{Zn})_2\text{O}_2$) – II a, b, c $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{AgNO}_3$ $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{ZnO}.$

Ag



4.

 $a \quad c$ b $x = 0,25.$ $x = 0,25$

(. 4).

Zn

 $a, b \quad c$ $= 0 - 0,12, \quad > 0,12$

(. 4).

(.).

[Cu]/[Li]

2

-

[O]/[Li]

 $2,2 - 2,3 = 2 +$ $= 0,2 - 0,3$

O ,

123,

[5].

Ag

,



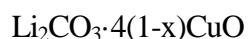
4 %. Ag

Ag

 $x = 0,25.$

Zn

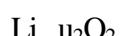
,



4xZnO = 0,10

12 %. (

).



4 %. Ag 12 %. Zn.

$(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$ 0 x 0,05 $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$ 0 x 0,12.

3.2.

LiCu₂O₂

() = M/H , (H 10),

40 ,

- (. 5).

,

-

,

, 1, Cu^{1+} ,

Cu^{2+}

$\mathbf{S} = 1/2$.

(T, H).

O ,

(T) $\text{DC} = 10$ 1 (. 5). < 400 K,

O , () (),

,

. 5 (T), ($H = 10$). ,

(T) 148 124,7 2 2, ,

(T) $H \ b$ 2

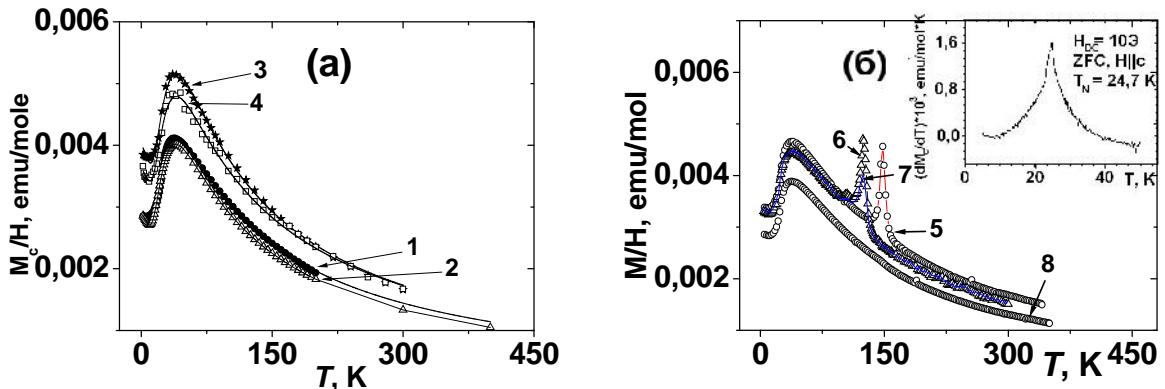
,

(FC, ZFC).

O

($H = 10$) = 150 $\text{LiCu}_2\text{O}_{2+}$

(. 5)



5. () \parallel DC = 10 , 1
 () (1 2) 1
 $f = 110$ $h_{ac} = 2$ (3,4);
 () 2 2 ZFC: 5,8-
 2 $H_{DC} = 10$, 6- 2
 918 , 1 $H_{DC} = 0,5$, 7- , $H_{DC} = 10$ (5,6,7
 $H||c$, 8- $H||b$ $H_{DC} = 10$).
 $dM_c(T)/dT$ 2, $N = 24,7$
 LiCu_2O_2 . c .

O
,
[6]
LiCu₂O₂ : CuO₆
,
[210].

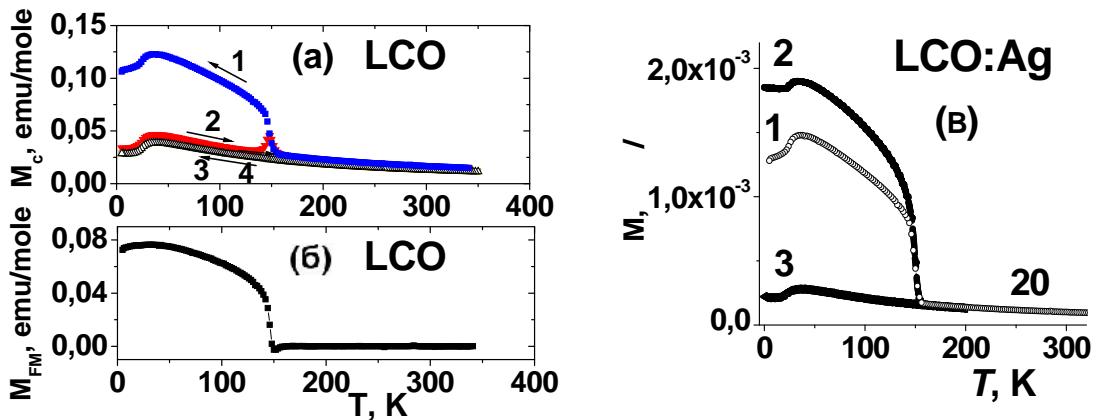
CuO_4 [7],

- [8]

, CuO_4

$$b - a (\dots , \dots , a \dots) -$$

b).



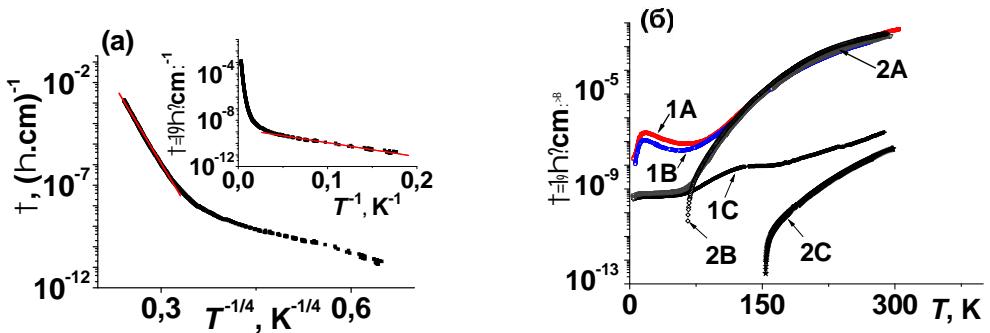
6.
 (1) FC; (2) ZFC; (4) $M_{FM} = M_{FC} - M_{ZFC}$
 10 : (3) $Li(Cu_{1-x}Ag_x)_2O_2$ = 0 (1), 0,05 (2) 0,15 (3)
 2. 20 .

LiCu₂O₂ .
 $Li_2O \cdot 4(1-x)CuO \cdot 4AgNO_3$ c 0 x 0,5, $M(T)$
 5-300 ($H=20$), **c**
 ZFC FC. ZFC FC $M(T)$
 $T = 37$ (. 6),
 $T_{c1} T_{c2}$ $dM(T)/dT$.
 $M(T)$ $LiCu_2O_2$,
 F $M(T)$ $x = 0,05$,
 $T_3 = 150$. T_3 ZFC FC.
 $Ag (x > 0,05)$
 $T_{c3} = 150$ (. 6).

3.3.

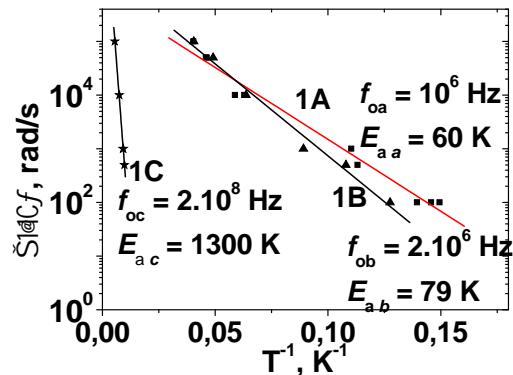
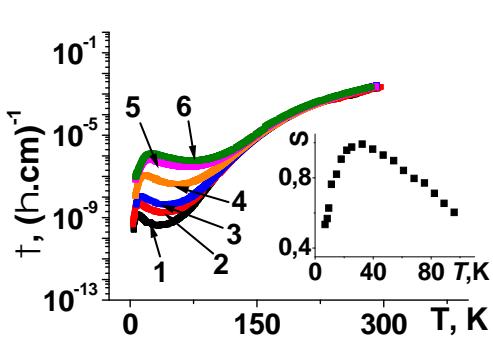
LiCu₂O₂.

$\sigma(=0) = \sigma_{DC}$ $LCO (. 8)$
 $T > 300$ K $\sigma_{DC} = \sigma_0 \exp(-E_a/kT)$
 E_a , 0,35 - 0,44 (), 100 - 300 -
 $\sigma_{DC} = \sigma_0 \exp(-T_0/T)^{1/4}$ $T_0 = 10^6 - 10^8$ K ().



$$7.) \quad \text{LiCu}_2\text{O}_2 \quad : \quad \begin{pmatrix} 1 & 1 & 1 \\ a, b, c; & 2 & - \end{pmatrix} \quad T^1 (\quad) ; \quad (0)$$

4,5



8. LiCu_2O_2 b (1, 2, 3, 4, 5 6
 $) (T,)$ 0,1, 0,5, 1, 10,50 100 $) s() ($).

9.

$T \sim 295$

6

M1

a = 1A, *b* = 1B, *c* = 1C

$$a \sim 5 - 6$$

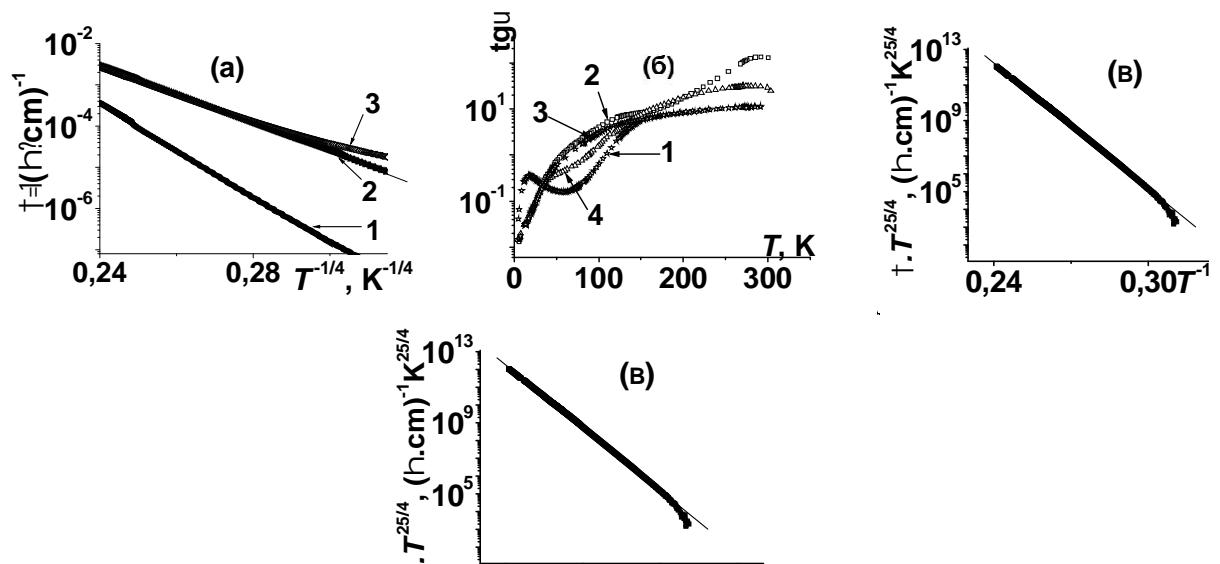
(. 8) -

03: 107 150

($\ln \sim 1/T$)

$$E_D \quad \quad \quad 0,15; 0,12 \quad 0,1$$

$$(\ln \sim T^{1/4}).$$



$\text{tg } (T)$ A4
 20 , ; 200

(. 10).

,
 , $\gg 1$

DC

ab

$\sim 10^{-10} (\cdot)^{-1}$ ~ 50 , ~ 100 .

(). , , *ab*

$m = 25/4$ $T_o = 6,75 \cdot 10^7$ (. 10),

[9], , , $\text{Li}^+ \text{ Cu}^{2+}$

,
 $\text{LiCu}_2\text{O}_{2+}$.

LiCu_2O_2

$\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$

$\text{Ag} \quad x = 0,15$

~ 3

$T < 240 \text{ K}$

$T \sim 25 \text{ K}$

$\text{AC}(\text{T})$

DC

$\text{Ag} (x = 0,05)$,

S-

(. . 11a),

LiCu_2O_2 [1].

$x > 0,05$

S

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4

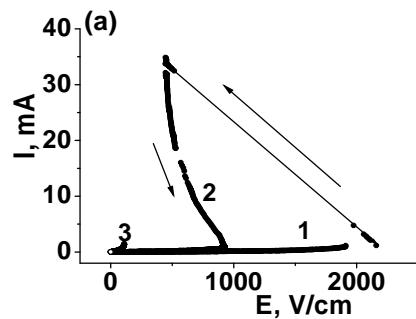
100 / (. . 11).

$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$

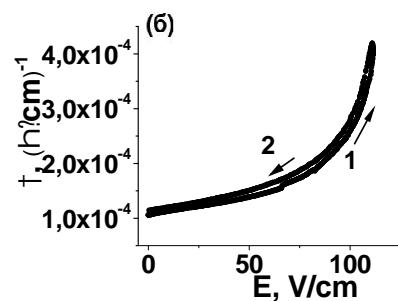
(. . 12).

Zn

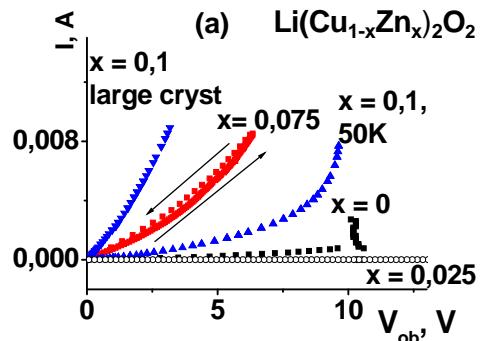
(. . 12).



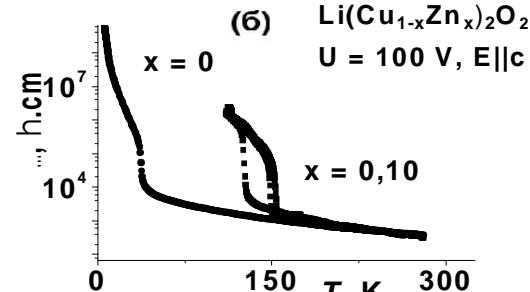
11.) ,
 $x = 0,15$ (3), (1, 2)
 $x = 0,15$ ().
 $x = 0,15$ (



$\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2 = 0,05$ (1, 2)
(3) c 78 .
, $\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$
ab 78).



12.) (T) $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$
 $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$,



100 .
1

1. LiCu_2O_2
2. LiCu_2O_2 ($\text{Li},\text{Ag})\text{Cu}_2\text{O}_2$, $\text{Li}(\text{Cu},\text{Zn})_2\text{O}_2$
 $3 \quad 10 \quad 10$
3. LCO
 $12 \quad .\% \text{ Zn (Cu) } 4\% \text{ Ag (Li).}$
4. $M(T)$
- O
 $\text{LiCu}_2\text{O}_2 \quad (H \quad 10 \quad) \quad = 150 \quad \mathbf{M} \parallel$
 O
 $\text{Ag} \quad (\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2 \quad (T < 50 \quad)$
 $x \quad 0,05$
 $150 \quad ,$
5. $\text{LCO} \quad 4,2 - 300 \quad 0,1 - 10,0$
 $5 \quad , \quad \sim 300$
 $(\quad = \quad \exp(-E_a/kT), E_a = 0,35 - 0,44 \quad)$

(= $\exp(-$
 $T_o/T^{1/4})$, $T_o = 10^6\text{-}10^8$), 25
 $E = 5$
 $- 6$

$$\lg \sim 1/T \quad \lg \sim T^{1/4}.$$

5 . LCO,

$$: a : b : c = 2 : 1 : 10^4. (295).$$

5 . - (T_f) tg (T_f),

$E_a = 60 \text{ - } 79$ K, $f_i = 10^6$ $E_a = 1300$ K, $f_i = 2 \cdot 10^8$

() (),

6. Ag Zn LiCu₂O₂

$$: x > 0,05 \quad \sim 3$$

$x < 0,05$

$$; \quad c x > 0,05$$

7. LCO

O

LCO , O ,

1 - 2 ,

O

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1. Wang K.F. Multiferroicity: the coupling between magnetic and polarization orders / K.F. Wang, J.M. Liu, Z.F. Ren // Adv. Phys. 2009. V. 58. - 04. - P. 321-448.

2. LiCu₂O₂ / . . . , . . .
 // . 2004. . 46. . 3. - . 433-440.

3. Roessli B. Magnetic Phase Transitions in the Double Spin-Chains Compound LiCu₂O₂ / B.

