

LiCu₂O₂,

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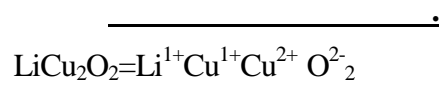
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www.misis.ru

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LCO

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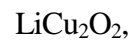
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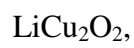
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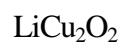
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- LiCu₂O₂
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- :
- Li₂CuO₂–CuO_x;
 - LiCu₂O₂
 Li(Cu,Zn)₂O₂, (Li,Ag)Cu₂O₂;
 4 10 10 ; LiCu₂O₂.
 - Li(Cu_{1-x}Zn_x)₂O₂, (Li_{1-x}Ag_x)Cu₂O₂
 $= 0 - 0,12$ $= 0 - 0,04$, Zn Ag
 LiCu₂O₂,
 Li(Cu_{1-x}Zn_x)₂O₂, (Li_{1-x}Ag_x)Cu₂O₂.
 - σ_{DC} ,
 $() = \sigma_{AC}$
 LiCu₂O₂ Li(Cu,Zn)₂O₂, (Li,Ag)Cu₂O₂ 4,2 – 300 0,1 – 100 .
 3 . , LiCu₂O₂ Li(Cu,Zn)₂O₂,
 (Li,Ag)Cu₂O₂ σ_{DC} $T \sim 300$ K
 - $(\sigma_{DC} = \sigma_0 \exp(E_a/k_B T))$
 (), $\sigma_{DC} = A \exp(T_0/T)^{1/4}$.
 ~25 DC

3 . LiCu_2O_2

$a, b, c: \dagger_a : \dagger_b : \dagger_c = 2 : 1 : 10^4$.

3 .

$\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$, $(\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$, $(x(\text{Zn}) = 0,05$,
 $x(\text{Ag}) < 0,02)$

S- - .

4. LCO

O_δ

LCO, O_δ ,

1 – 2 ,

150 K. O_δ

_____.

LiCu_2O_2

$\text{Li}(\text{Cu},\text{Zn})_2\text{O}_2$, $(\text{Li},\text{Ag})\text{Cu}_2\text{O}_2$:

LiCu_2O_2

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 ,
 CuO₄-
 (ladder compound),
 ,

LiCu₂O₂

LiCu₂O₂

: 1) -Cu¹⁺(1)-, 2) -O(1)Cu²⁺(2)O(2)Li- 3) -

LiO(2)Cu²⁺(2)O(1) [1] (. 1). Cu¹⁺ ,

O²⁻-Cu¹⁺-O²⁻ , LiCuO₂- . 2) 3)

CuO₅ LiO₅, **ab**-

LiO₄ CuO₄ , **a** ,

Cu-O- Li-O- , **b** -

Cu-O- Li-O- . 1) Cu¹⁺ .

LiCu₂O₂ -

Cu²⁺-O , -LiCuO₂- ,

, **b**.

-

Li¹⁺ Cu¹⁺ ,

.

Cu-O- ,

[1]). LiCu₂O₂

.

$T_{c1} = 24,6$ T_{c2}

$= 23,2$:

T_{c1} T_{c2} T_{c2} [1].

c-

P_s. T_{c2} ,

[1].

LiCu₂O₂ ,

1) Cu¹⁺ ,

- O²⁻, p- (. 1).

O²⁻

Cⁿ⁺ = Cu²⁺ Li¹⁺ ,

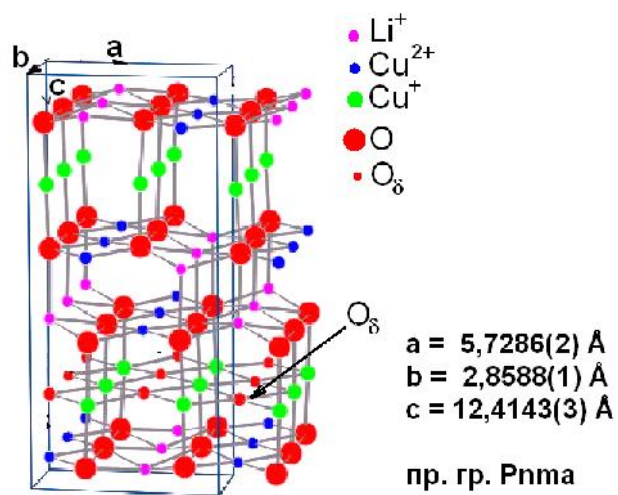
CO₅ CO₆.

O2p-, Cu3d-O2p- CuO₄.

Pmna

[3]. ,

LCO - Cu²⁺



1. LiCu_2O_2 ([1]).

Li^+ $\text{u}(\text{Li})\text{-O-Cu}(\text{Li})$, - LCO.

LiCu_2O_2 ,

LiCu_2O_2 $(\text{Li},\text{Na})\text{Cu}_2\text{O}_2$ [4],

$\text{Li}(\text{Cu},\text{Zn})_2\text{O}_2$ $\text{Li}(\text{Cu},\text{Ni})_2\text{O}_2$ [5].

LiCu_2O_2

,

2)

-2-

()

-4

: $(\text{CuK}_1) = 1,54051$, $(\text{CuK}_2) = 1,54433$, $(\text{CuK}_{\text{av}}) = 1,54178 \text{ \AA}$.

“Orbis” “EDAX” () Si(Li)
 , Na.
 () Q1500
 D F. Paulik, J. Paulic, L. Erdey.

7-20 100 – 100 ,
 ,
 $R_1 = 5,26$,
 5 – 300 SQUID
 MPMS-XL-7 Quantum Design Inc ($H = 20$)
 (zero- field-cooled – ZFC) (field-cooled – FC).

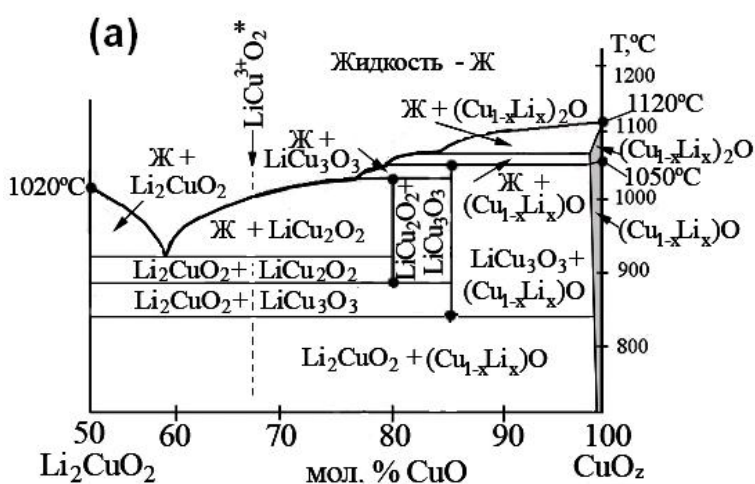
LiCu₂O₂ ,
 .

3.1. , Li_2CuO_2 - CuO .

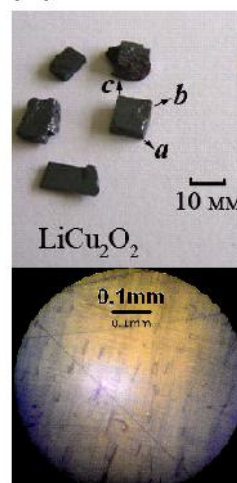
(. 2).
 : LiCu₂O₂ LiCu₃O₃,
 (1323 1373) (1163 1113
).

1173 - 1323

$LiCu_2O_2$.
 CuO « . ». Li₂CO₃ « . ».
 LiCu₂O₂ xCuO·(1-x)Li₂CO₃ c 0,77 x 0,83 4 1393 0,5
 , 1323 ,
 2,0 / - 1173 ,
 1173 10 - 20



(б)



2.) $\text{Li}_2\text{CuO}_2 - \text{CuO}_z$
 ;) LiCu_2O_2 ()
 LiCu_2O_2 , () (001).

1393

1323

~1173

 LiCu_2O_2 .

LiCu_2O_2 , (0,5 – 4) $8 \times 8 \times 10^{-3}$ (. 2).
 (001),

(210) (2-10),

{001}

{210}.

 LiCu_2O_2 ,

(120) (1-20) (. 2).

 LiCu_2O_2 LiCu_2O_2

Ag

Zn

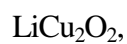
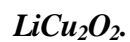
LCO.

$\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{AgNO}_3$, $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{ZnO}$ - I (1-x) $\text{Li}_2\text{CO}_3 \cdot 2x\text{ZnO} \cdot 4\text{CuO}$ - II 0 x 0,5,

 Li_2CO_3 , CuO, AgNO₃, ZnO

« », « », « » « »

, LCO,
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> 0,15 1 .



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5,0 / .

6 20 . *a*

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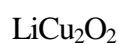
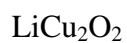
~1113

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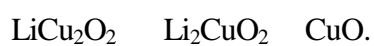
400

1 - 4 .

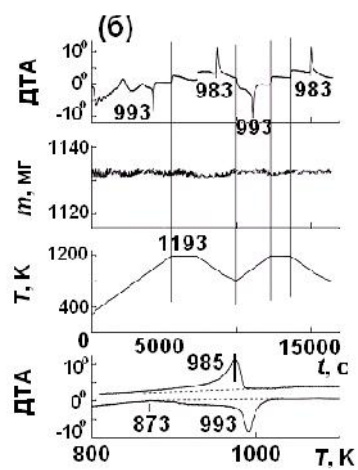
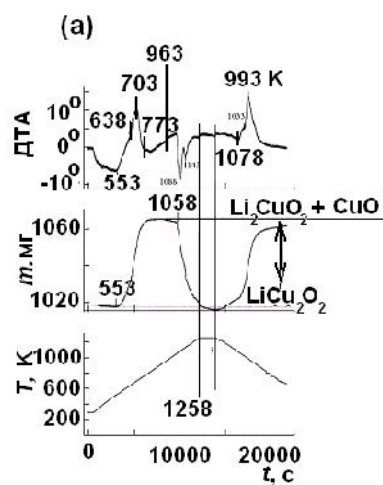


[3],

$T = 553-773$



m



3.

$LiCu_2O_2$:)

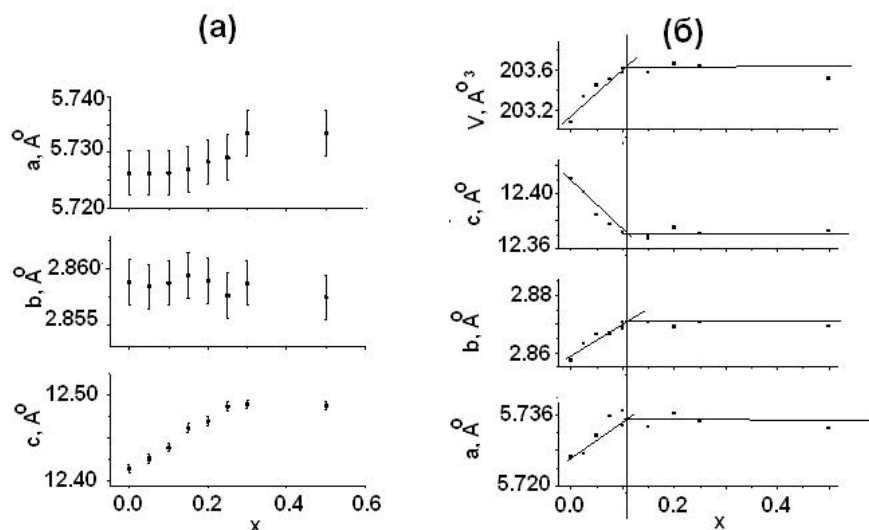
)

(. 3) Cu^+ Cu^{2+} ,
 $\text{LiCu}^{2+}\text{Cu}^+\text{O}_2$.
1073–1223 LiCu_3O_3 ,
 LiCu_2O_2 , .
 $\text{Li}(\text{Cu}_{0,95}\text{Zn}_{0,05})_2\text{O}_2$ LiCu_2O_2 .

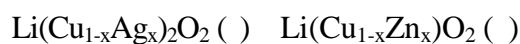
LiCu_2O_2			
1	-		2 400
2	-		-
1			-
2			-
A4	-	40 . 1113	2 4 400
W2	-	40 . 1113	

LiCu_2O_2 LiCu_2O_2 .
 LiCu_2O_2 (1320). $T_{\text{PT}} = 993$
, (. 3).
,
 . $T = 983$
, .
 LiCu_2O_2 .

: $a = 5,73$, $b = 2,86$, $c = 12,41 \text{ \AA}$,
[1] LiCu_2O_2 .
 ZnO , Zn
 LiCu_2O_2 ((Li,Zn) Cu_2O_2 - I, $\text{Li}(\text{Cu,Zn})_2\text{O}_2$ – II. II.
 . 4 a, b, c
 $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{AgNO}_3$
 $\text{Li}_2\text{CO}_3 \cdot 4(1-x)\text{CuO} \cdot 4x\text{ZnO}$. Ag



4.

 a, c b $x = 0,25.$ $x = 0,25$

(. 4).

Zn

 a, b, c $= 0 - 0,12, > 0,12$

(. 4).

-

().

-

 $[\text{Cu}]/[\text{Li}]$

,

2

-

,

 $[\text{O}]/[\text{Li}]$ $2,2 - 2,3 = 2 +$

,

 $= 0,2 - 0,3$

O ,

,

123,

[5].

,

Ag

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4 . %

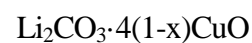
Ag

 $x = 0,25.$

.

Zn

,

 $4x\text{ZnO} = 0,10$

12 . % (

).

,

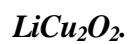
,

 $\text{Li}_2\text{u}_2\text{O}_2$

4 . % Ag 12 . % Zn.



3.2.



$$\langle \sigma \rangle = M/H, \quad (H = 10^{-10}),$$

40 ,

- (. 5).

,

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—

;

 Cu^{1+} ,

1,

$$\text{Cu}^+$$

Cu^{2+}

$$S = 1/2.$$

$(T, H).$

0,

.

$$(T) \quad \begin{array}{lll} \text{DC} = 10 & 1 \text{ (} \cdot 5 \text{)}. & < 400 \text{ K,} \\ \text{O, (} & & \text{ ()}, \end{array}$$

• ,

•

5 (T), (H = 10). ,

•

(T)	148	124,7	2	2,	,
		(T)	H	b	2

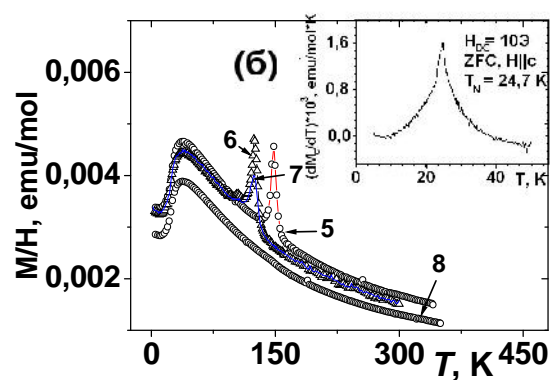
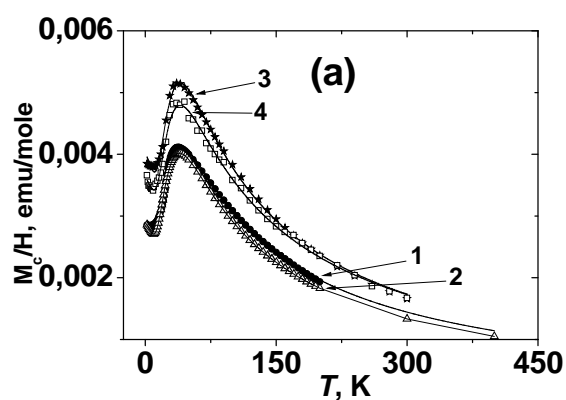
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(FC, ZFC).

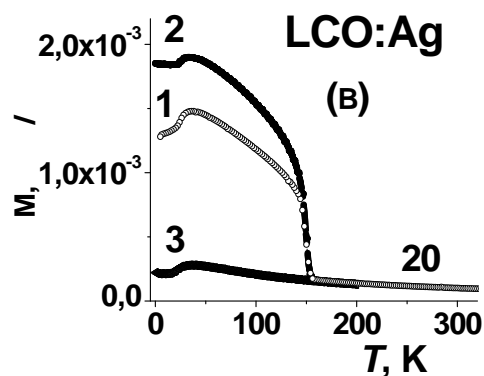
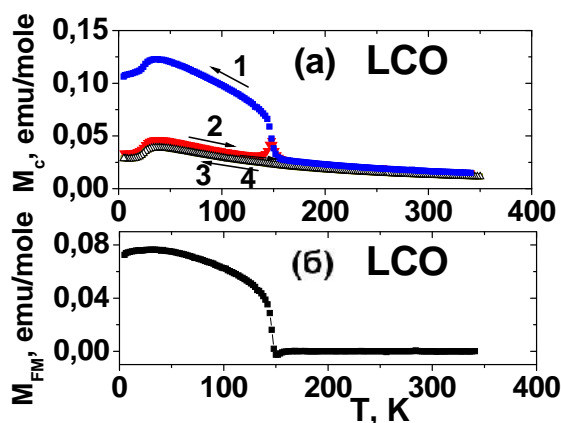
O

$$(H = 10) = 150 \text{ LiCu}_2\text{O}_{2+} \quad (0.5)$$



5. () \parallel $H_{DC} = 10$, () :
 (1 2) $f = 110$ $h_{ac} = 2$ (3, 4) ;
 () $H_{DC} = 10$, $6 -$ 2 ZFC: 5, 8 -
 918 , 1 $H_{DC} = 0,5$, $7 -$, $H_{DC} = 10$ (5, 6, 7
 $H \parallel c$, $8 -$ $H \parallel b$ $H_{DC} = 10$).
 $dM_c(T)/dT$ 2, $N = 24,7$
 $LiCu_2O_2$. **c.**

·
 O
 ,
 [6] -
 $LiCu_2O_2$: CuO_6
 ,
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 [210].
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 CuO_4 [7],
 - [8]
 +1.
 , CuO_4
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 . 6() , **c**
 b a (, a -
 b).

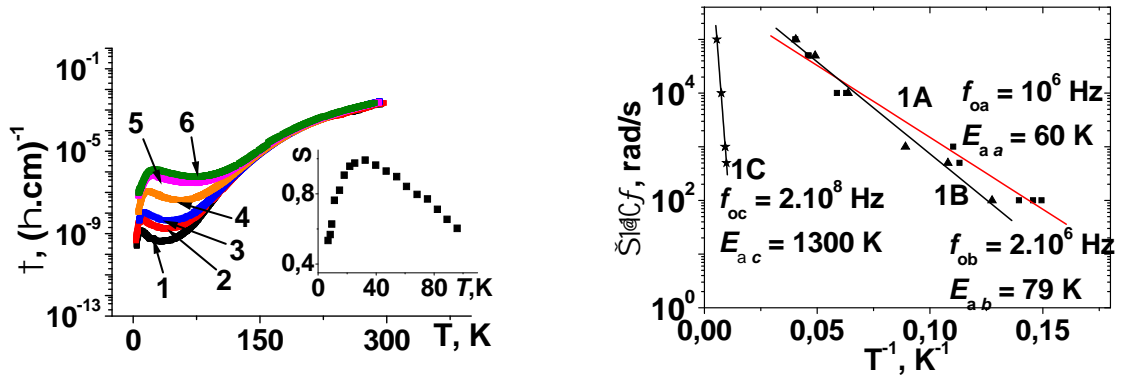
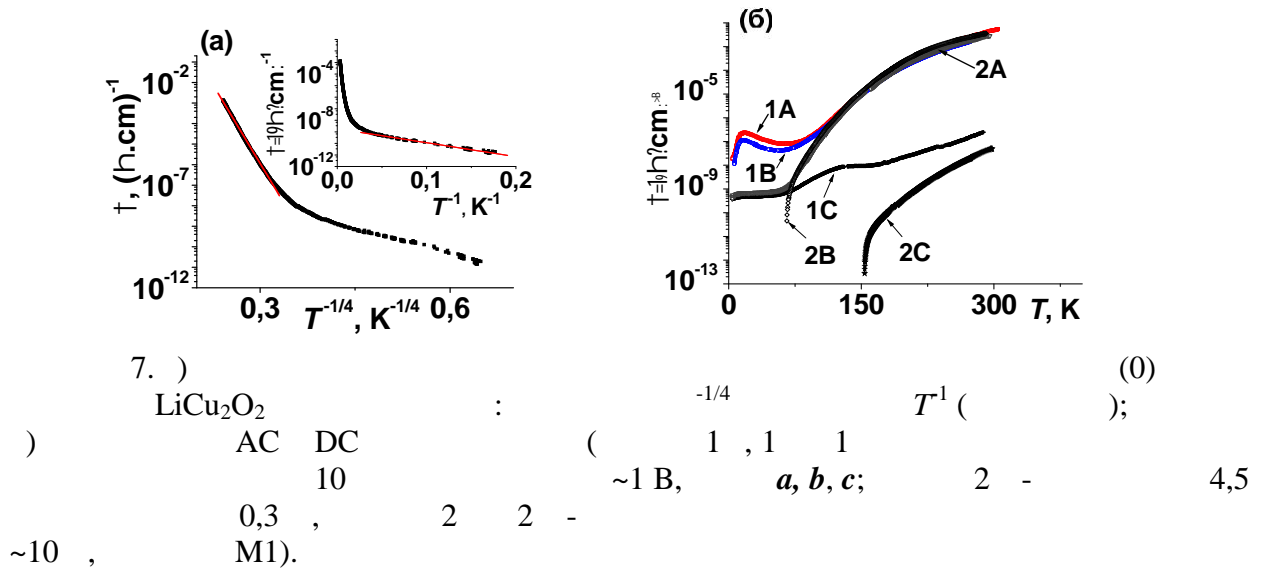


6. LiCu_2O_2 () ($H = c$, $H_{\text{DC}} = 10$:
 (1) FC; (2) ZFC; (3) ZFC; (4) FC.)
 10 : (3) $M_{\text{FM}} = M_{\text{FC}} - M_{\text{ZFC}}$ ()
 2. $\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ = 0 (1), 0,05 (2) 0,15 (3)
 () , 20 .

LiCu_2O_2 .
 $\text{Li}_2\text{O} \cdot 4(1-x)\text{CuO} \cdot 4\text{AgNO}_3$ c 0 $x = 0,5$, $M(T)$
 5–300 (H=20), **c**
 ZFC FC. ZFC FC $M(T)$
 $T = 37$ (. 6),
 $T_{\text{c1}} \quad T_{\text{c2}} \quad dM(T)/dT$.
 $M(T)$ LiCu_2O_2 .
 F $M(T)$ $x = 0,05$, ,
 $T_3 = 150$. T_3 ZFC FC.
 Ag ($x > 0,05$)
 $T_{\text{c3}} = 150$ (. 6).

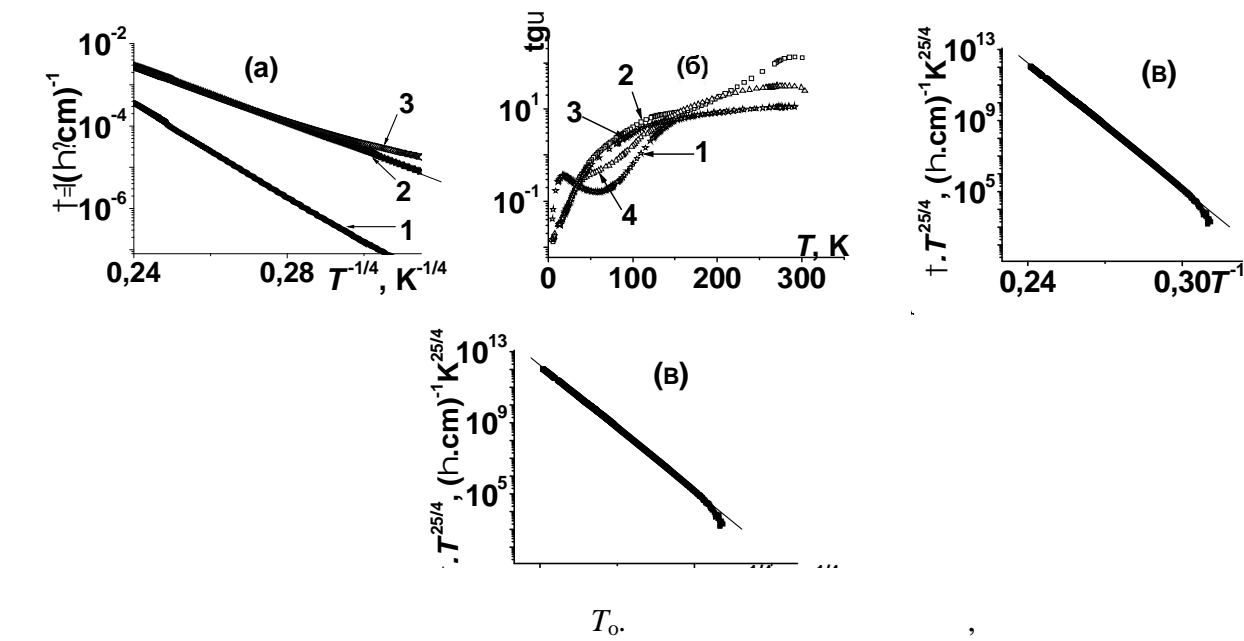
3.3.

LiCu_2O_2 .
 $\sigma(\omega=0) = \sigma_{\text{DC}}$ LCO (. 8)
 $T > 300$ K $\sigma_{\text{DC}} = \sigma_0 \exp(-E_a/kT)$
 E_a , 0,35 - 0,44 (), 100 - 300 -
 $\sigma_{\text{DC}} = \sigma_0 \exp(-T_0/T)^{1/4}$ $T_0 = 10^6 - 10^8$ K ().
 25 DC :



8.) (T,) LiCu_2O_2 b (1, 2, 3, 4, 5 6
 0,1, 0,5, 1, 10,50 100) s() ().
 9. M1 a – 1A, b – 1B, c – 1C.
 a ~5 - 6 ,
 (. 8) –
 T ~295 c ab,
 b a.
 $E_D = -d(\ln)/d(1/T)$,
 ~200 E_D 0,15; 0,12 0,1
 0,3; 107 150 ,
 (ln ~ 1/T) (ln ~ $T^{1/4}$).

$T < 30$ ***ab***,
 , ***a*** ***b***
16,6 15,7 , ***c***
- ~134 .
 .
. 8 (T ,), M1 ***b***
 , ~100
80 , 80
 , (T ,) = T^n s ~1,
()
 .
 $s(T)$ () = 30 ,
 .
 .
 , $\text{tg } (T)$
a, ***b***, ***c***, , $\lg f - 1/T_{\max}$,
(. 9).
($= 1/2 f$), =
 $_{\text{oi}}\exp(E_{\text{ai}}/kT)$, $i = \textbf{\textit{a}}, \textbf{\textit{b}}, \textbf{\textit{c}}$. f_{oi} ,
 ,
 $\sim 10^6$ ***a*** ***b***, $2 \cdot 10^8 -$. $f_{\text{oa,b}}$
(), $E_{\text{a} \textbf{\textit{a}}, \textbf{\textit{b}}}$
 CuO_2 (***b***) () [1].
c f_o
(). , $< 30 \text{ K}$
 , $> 30 \text{ K} -$.
 .
DC 295 ***ab*** 20%, ***c*** -
 .
 - (),
 , ***c*** , , ,
 .
 T_o
(. 10).



$\text{tg } (T)$ A4

20 , ; 200

(. 10).

, ,

$\gg 1$

DC

ab

$\sim 10^{-10} (\cdot)^{-1}$ ~50 , - ~100 . ,

(). , ab

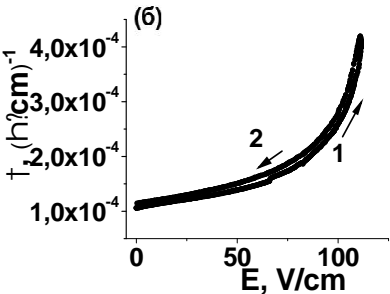
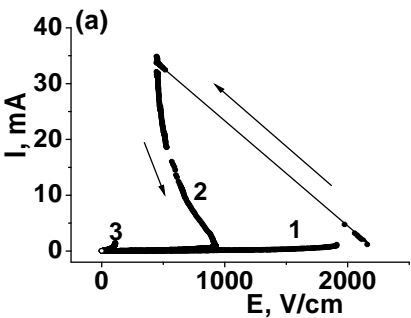
$m = 25/4 \quad T_0 = 6,75 \cdot 10^7 (\cdot 10),$

[9], , , $\text{Li}^+ \text{Cu}^{2+}$

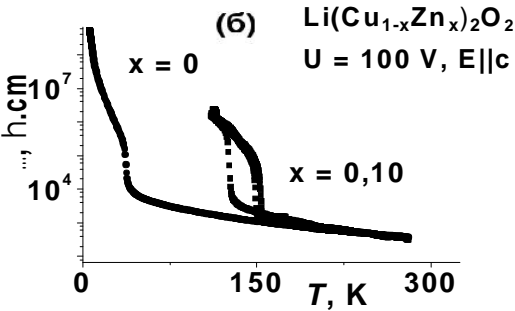
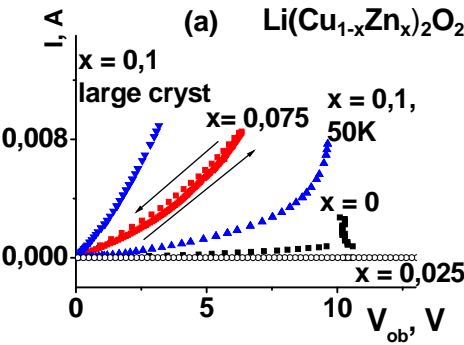
,
 $\text{LiCu}_2\text{O}_{2+}$.

LiCu_2O_2 . ,

$\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ Ag $x = 0,15$
~3 . DC
 $T < 240 \text{ K}$.
 $T \sim 25 \text{ K}$ $\sigma_{AC}(T)$.
, Ag ($x = 0,05$), S-
(. 11a),
 LiCu_2O_2 [1]. , $x > 0,05$
, S , ,
:
4 $100 /$ (. 11).
 $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$ (. 12).
Zn ,
(. 12).



11.) ,
0,15 (3), (1, 2)
) .
=0,15 ($\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$ = 0,05 (1, 2)
(3) c 78 .
, $\text{Li}(\text{Cu}_{1-x}\text{Ag}_x)_2\text{O}_2$
 ab 78).



12.) $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$;
(T) $\text{Li}(\text{Cu}_{1-x}\text{Zn}_x)_2\text{O}_2$, 1
100 .

1. LiCu_2O_2
2. $\text{Li}_2\text{CuO}_2\text{-CuO}_x$
3. LiCu_2O_2 $(\text{Li,Ag})\text{Cu}_2\text{O}_2$, $\text{Li}(\text{Cu,Zn})_2\text{O}_2$
4. $M(T)$, LCO
5. LiCu_2O_2 $(H = 10 \text{ Oe})$ $= 150$ $\text{M} \parallel$
6. $\text{Ag} (\text{Li}_{1-x}\text{Ag}_x)\text{Cu}_2\text{O}_2$ $(T < 50 \text{ Oe})$
7. $x = 0,05$
8. 150
9. LCO $4,2 - 300$ $0,1 - 10,0$
10. ~ 300
11. $(= \exp(-E_a/kT), E_a = 0,35 - 0,44)$

($= \exp(-E/kT)$), $T_0 = 10^6 - 10^8$ K, $E = 5 \cdot 10^{-6}$ eV, 25

$\lg \sim 1/T$, $\lg \sim T^{1/4}$.
 5. LCO, $a:b:c = 2:1:10^4$ (295 K).
 5. $(T_i f) \sim \lg(T_i f)$,
 ,

$E_a = 60 - 79$ K, $f_i = 10^6$ $E_a = 1300$ K, $f_i = 2 \cdot 10^8$
 (), (),

6. Ag Zn LiCu_2O_2
 $x < 0,05$ $x > 0,05$ ~ 3
 $c x > 0.05$

7. LCO
 ,
 O
 LCO , O,
 ,
 1 - 2 ,
 O ,

1. Wang K.F. Multiferroicity: the coupling between magnetic and polarization orders / K.F. Wang, J.M. Liu, Z.F. Ren // Adv. Phys. 2009. V. 58. — 04. — P. 321–448.

2. . . LiCu_2O_2 / . . . , . . .
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Roessli, U. Staub, A. Amato [] // Physica B. 2001. V. 296. – P.306–311.

4. LiCu₂O₂-NaCu₂O₂ /
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